

TCEQ Class B Surface Water License Practice Test (Sample)

Study Guide



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SAMPLE

Questions

- 1. What advantages does a Wagner block provide in filtration systems?**
 - A. High cost and complexity**
 - B. Simplicity and reduced gravel needs**
 - C. Increased weight stability**
 - D. Enhanced filtration speed**
- 2. How many years are sludge regulations permitting sludge to be kept on site before disposal?**
 - A. 1 year**
 - B. 2 years**
 - C. 3 years**
 - D. 5 years**
- 3. What three purposes do under-drains serve?**
 - A. They stabilize the ground**
 - B. They enhance water pressure**
 - C. They support the filter media**
 - D. They prevent soil erosion**
- 4. What is the most important factor affecting the useful life of service lines?**
 - A. Abundance of water supply**
 - B. Ability of material to resist internal and external corrosion**
 - C. Type of pump used in the service**
 - D. Diameter of the service line**
- 5. What is formed when all chlorine demand is met?**
 - A. Chlorine residual**
 - B. Chlorine gas**
 - C. Chlorine demand**
 - D. Chlorinated organics**

- 6. How does tuberculation affect a pipe's C factor?**
- A. It increases the C factor**
 - B. It has no effect on the C factor**
 - C. It decreases the C factor**
 - D. It doubles the C factor**
- 7. What is the maximum sustained gas withdrawal rate for 150 lbs chlorine cylinders?**
- A. 1 lb per day per degree F**
 - B. 4 lbs per day per degree F**
 - C. 8 lbs per day per degree F**
 - D. 10 lbs per day per degree F**
- 8. What is a standard solution?**
- A. A solution with an unknown strength**
 - B. A solution of known strength or concentration**
 - C. A solution used for emergency purposes**
 - D. A solution used in filtration processes**
- 9. Should filter sidewalls be smooth to avoid short-circuiting between the walls and sand?**
- A. Yes, they should be smooth**
 - B. No, they should be roughened**
 - C. It depends on the design**
 - D. Only for certain types of filters**
- 10. What fluoride level is considered moderate and beneficial for dental health?**
- A. 0.5 mg/l**
 - B. 0.7 to 1.0 mg/l**
 - C. 1.5 mg/l**
 - D. 2.5 mg/l**

Answers

SAMPLE

- 1. B**
- 2. B**
- 3. C**
- 4. B**
- 5. A**
- 6. C**
- 7. A**
- 8. B**
- 9. B**
- 10. B**

SAMPLE

Explanations

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1. What advantages does a Wagner block provide in filtration systems?

- A. High cost and complexity**
- B. Simplicity and reduced gravel needs**
- C. Increased weight stability**
- D. Enhanced filtration speed**

A Wagner block is specifically designed to enhance the efficiency and effectiveness of filtration systems by simplifying the configuration required for effective filtering. One of the key advantages of a Wagner block is its ability to reduce the amount of gravel needed in the filtration process. This is significant because it not only lowers material costs but also simplifies the setup and maintenance of the system. By using a Wagner block, operators can achieve a high level of filtration without the need for extensive gravel layers, which can be cumbersome to install and maintain. The design of the block allows for better management of water flow and filtration, making it a practical choice for various water treatment applications. Overall, the simplicity and reduction in gravel needs position the Wagner block as a cost-effective and efficient solution in filtration systems.

2. How many years are sludge regulations permitting sludge to be kept on site before disposal?

- A. 1 year**
- B. 2 years**
- C. 3 years**
- D. 5 years**

The regulations for sludge management established by the Environmental Protection Agency (EPA) generally permit sludge to be stored on site for a maximum of two years prior to its disposal. This two-year timeframe is intended to ensure that sludge is properly managed and does not pose a risk to public health or the environment. Proper storage practices during this period are crucial for preventing odors, vector attraction, and the leaching of contaminants. While some facilities may choose to dispose of sludge sooner based on operational requirements or capacity, the regulation clearly stipulates that, under normal circumstances, a two-year limit is imposed. This regulation helps maintain a balance between effective waste management and environmental protection, ensuring that the contents of the sludge do not degrade to a point that would create additional issues for disposal or treatment processes. It's important to also understand that the other options represent timeframes that either exceed or do not align with the regulatory requirements for sludge storage, reinforcing the significance of adhering to the two-year guideline.

3. What three purposes do under-drains serve?

- A. They stabilize the ground
- B. They enhance water pressure
- C. They support the filter media**
- D. They prevent soil erosion

Under-drains primarily serve to support the filter media in a drainage system. This function is critical because the filter media is designed to separate sediments and pollutants from the water while facilitating drainage. If the filter media is properly supported, it maintains its structure and efficiency, ensuring that water can pass through effectively without clogging. Additionally, under-drains provide a pathway for excess water to flow away, which helps maintain the proper moisture balance within the surrounding soil. When soil becomes overly saturated, it can lead to a reduction in the effectiveness of the filter media, which could compromise water quality. Thus, the role of under-drains in supporting filter media is fundamental for both the operational efficiency of the drainage system and the protection of water quality in the surrounding environment.

4. What is the most important factor affecting the useful life of service lines?

- A. Abundance of water supply
- B. Ability of material to resist internal and external corrosion**
- C. Type of pump used in the service
- D. Diameter of the service line

The most important factor affecting the useful life of service lines is the ability of the material to resist internal and external corrosion. Corrosion can significantly reduce the structural integrity and lifespan of service lines, which are typically made from materials like metal or plastic. When a service line is subjected to corrosive conditions—whether from the water itself, environmental factors, or other external sources—the material can deteriorate over time, leading to leaks, failures, or contamination of the water supply. While factors such as water supply abundance, pump type, and service line diameter may influence the operational efficiency or flow capacity, they do not directly impact the durability of the materials used in the construction of the lines. Understanding how different materials stand up to corrosion is critical in ensuring the longevity and reliability of water service infrastructure. This is why emphasis is placed on selecting and maintaining materials that provide strong corrosion resistance, as it is a key determinant in the overall longevity and safety of service lines.

5. What is formed when all chlorine demand is met?

- A. Chlorine residual**
- B. Chlorine gas**
- C. Chlorine demand**
- D. Chlorinated organics**

When all chlorine demand is met, a chlorine residual is formed. Chlorine demand refers to the amount of chlorine that is required to react with various substances in the water, such as organic matter, bacteria, and other compounds that can consume chlorine. Once all of these demand reactions have taken place and the chlorine has successfully interacted with those impurities, any remaining chlorine that does not get used up is known as the chlorine residual. This residual is important because it indicates the amount of chlorine that is still present in the water to continue disinfecting and protecting against the regrowth of microorganisms as the water is distributed through the system. Maintaining an adequate level of chlorine residual is critical for effective water treatment. It serves as a measure of the efficacy of the chlorination process and ensures that the water remains safe for consumption. Chlorine gas, chlorine demand, and chlorinated organics represent different concepts in the context of water treatment and do not apply to the requirement of having an unspent amount of chlorine that stays in the system.

6. How does tuberculation affect a pipe's C factor?

- A. It increases the C factor**
- B. It has no effect on the C factor**
- C. It decreases the C factor**
- D. It doubles the C factor**

Tuberculation refers to the formation of small, protruding deposits on the interior surface of pipes, particularly in cast iron or ductile iron systems. Over time, this buildup can occur due to the corrosion and reaction of water with the pipe material, leading to rough, uneven surfaces. The C factor, or Hazen-Williams coefficient, is a measure of the roughness of a pipe's interior surface. A higher C factor indicates a smoother pipe, which allows water to flow more freely, while a lower C factor signifies a rougher surface that creates more friction and resistance to flow. When tuberculation occurs, the interior surfaces of the pipe become rougher, leading to increased turbulence and friction for the flowing water. As a result, the C factor decreases because the energy loss due to friction in a rough pipe is greater compared to a smooth pipe. Therefore, the presence of tuberculation negatively impacts the hydraulic performance of the pipe by reducing its flow capacity. Understanding tuberculation's effect on the C factor is crucial for maintaining the efficiency of water distribution systems, as it helps in estimating pressure losses and flow rates in pipelines. This knowledge is essential for anyone involved in water management, pipe maintenance, or engineering design in the context of surface

7. What is the maximum sustained gas withdrawal rate for 150 lbs chlorine cylinders?

- A. 1 lb per day per degree F**
- B. 4 lbs per day per degree F**
- C. 8 lbs per day per degree F**
- D. 10 lbs per day per degree F**

The maximum sustained gas withdrawal rate for a 150 lb chlorine cylinder is 1 lb per day per degree Fahrenheit. This rate is established based on safety and operational standards for handling chlorine gas, ensuring that the gas can be withdrawn safely without exceeding the cylinder's capacity and maintaining proper pressure levels. This limit helps to prevent issues such as the liquid phase of chlorine entering the withdrawal system, which can lead to equipment damage, inefficient dosing, and potential safety hazards. By adhering to this guideline, operators can ensure that they manage chlorine use effectively and maintain safe operational practices. Proper understanding and adherence to these rates are crucial for facility operators and technicians when managing chemical feeds in water treatment processes.

8. What is a standard solution?

- A. A solution with an unknown strength**
- B. A solution of known strength or concentration**
- C. A solution used for emergency purposes**
- D. A solution used in filtration processes**

A standard solution is defined as a solution of known strength or concentration. It is used in various analytical chemistry processes, specifically in titrations, where the precise concentration of a reactant is crucial for achieving accurate and reliable results. By knowing the exact concentration of a standard solution, a chemist can determine the concentration of an unknown solution through comparison and calculation. This precise knowledge allows for various quantitative analyses, such as determining the amount of a substance in a sample or calibrating instruments used in measurements. The accuracy of the analysis heavily relies on the ability to work with a solution that has a well-defined concentration, making the standard solution a foundational element in laboratory practices. Other types of solutions, such as those with unknown strengths or used for emergency or filtration purposes, lack this defined concentration and therefore do not serve the same purpose in analytical contexts. The clarity and reliability provided by a standard solution are what make it essential in chemical analysis and research.

9. Should filter sidewalls be smooth to avoid short-circuiting between the walls and sand?

- A. Yes, they should be smooth**
- B. No, they should be roughened**
- C. It depends on the design**
- D. Only for certain types of filters**

The correct understanding regarding filter sidewalls is that they should actually be smooth to minimize the risk of short-circuiting between the walls and the sand. Smooth walls promote even flow distribution and help ensure that water travels through the filter media effectively, rather than bypassing the filter where it could flow directly from one side to another without proper treatment. Rough walls, on the other hand, may create turbulence that can lead to uneven flow patterns. This roughness could potentially disrupt the filtering process by allowing water to take shortcuts, thus reducing the filtration efficiency. Proper design considerations for filters emphasize the importance of smooth surfaces to maintain the integrity of the filtration process and ensure that contaminants are trapped effectively within the filter media. It's also crucial to note that some variable factors in filter design do exist, but when focusing specifically on the characteristics of the sidewalls, smooth surfaces are preferred to provide optimal filtration performance.

10. What fluoride level is considered moderate and beneficial for dental health?

- A. 0.5 mg/l**
- B. 0.7 to 1.0 mg/l**
- C. 1.5 mg/l**
- D. 2.5 mg/l**

The fluoride level considered moderate and beneficial for dental health typically falls within the range of 0.7 to 1.0 mg/l. This concentration is widely recognized and supported by dental health organizations as optimal for reducing the incidence of dental caries (cavities) while minimizing the risk of dental fluorosis, which is a condition that can occur from excessive fluoride exposure during tooth development. At this level, fluoride is effective in enhancing the remineralization of tooth enamel and can inhibit the growth of harmful oral bacteria. This range strikes a balance where fluoride provides protective benefits for teeth without leading to adverse effects, making it a well-established guideline for fluoride concentration in community water supplies for promoting oral health.