

# Natural Science CLEP Prep Practice Exam (Sample)

## Study Guide



**Everything you need from our exam experts!**

**This is a sample study guide. To access the full version with hundreds of questions,**

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# Introduction

Preparing for a certification exam can feel overwhelming, but with the right tools, it becomes an opportunity to build confidence, sharpen your skills, and move one step closer to your goals. At Examzify, we believe that effective exam preparation isn't just about memorization, it's about understanding the material, identifying knowledge gaps, and building the test-taking strategies that lead to success.

This guide was designed to help you do exactly that.

Whether you're preparing for a licensing exam, professional certification, or entry-level qualification, this book offers structured practice to reinforce key concepts. You'll find a wide range of multiple-choice questions, each followed by clear explanations to help you understand not just the right answer, but why it's correct.

The content in this guide is based on real-world exam objectives and aligned with the types of questions and topics commonly found on official tests. It's ideal for learners who want to:

- Practice answering questions under realistic conditions,
- Improve accuracy and speed,
- Review explanations to strengthen weak areas, and
- Approach the exam with greater confidence.

We recommend using this book not as a stand-alone study tool, but alongside other resources like flashcards, textbooks, or hands-on training. For best results, we recommend working through each question, reflecting on the explanation provided, and revisiting the topics that challenge you most.

Remember: successful test preparation isn't about getting every question right the first time, it's about learning from your mistakes and improving over time. Stay focused, trust the process, and know that every page you turn brings you closer to success.

Let's begin.

# How to Use This Guide

**This guide is designed to help you study more effectively and approach your exam with confidence. Whether you're reviewing for the first time or doing a final refresh, here's how to get the most out of your Examzify study guide:**

## **1. Start with a Diagnostic Review**

**Skim through the questions to get a sense of what you know and what you need to focus on. Don't worry about getting everything right, your goal is to identify knowledge gaps early.**

## **2. Study in Short, Focused Sessions**

**Break your study time into manageable blocks (e.g. 30 - 45 minutes). Review a handful of questions, reflect on the explanations, and take breaks to retain information better.**

## **3. Learn from the Explanations**

**After answering a question, always read the explanation, even if you got it right. It reinforces key points, corrects misunderstandings, and teaches subtle distinctions between similar answers.**

## **4. Track Your Progress**

**Use bookmarks or notes (if reading digitally) to mark difficult questions. Revisit these regularly and track improvements over time.**

## **5. Simulate the Real Exam**

**Once you're comfortable, try taking a full set of questions without pausing. Set a timer and simulate test-day conditions to build confidence and time management skills.**

## **6. Repeat and Review**

**Don't just study once, repetition builds retention. Re-attempt questions after a few days and revisit explanations to reinforce learning.**

## **7. Use Other Tools**

**Pair this guide with other Examzify tools like flashcards, and digital practice tests to strengthen your preparation across formats.**

**There's no single right way to study, but consistent, thoughtful effort always wins. Use this guide flexibly — adapt the tips above to fit your pace and learning style. You've got this!**

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## Questions

- 1. What is the scientific name for the study of energy and matter?**
  - A. Biology**
  - B. Chemistry**
  - C. Physics**
  - D. Psychology**
- 2. What is the study of the structure and behavior of systems made up of many interacting parts?**
  - A. Ecology**
  - B. Economics**
  - C. Thermodynamics**
  - D. Systems Theory**
- 3. What is the smallest representative unit in chemistry?**
  - A. Atom**
  - B. Molecule**
  - C. Ion**
  - D. Proton**
- 4. What is the acronym for the study of living organisms and their environment?**
  - A. Biodiversity**
  - B. Ecology**
  - C. Earth Science**
  - D. Environmental Science**
- 5. What is the most abundant element in the Earth's crust?**
  - A. Oxygen**
  - B. Iron**
  - C. Carbon**
  - D. Hydrogen**



- 6. What is the study of the chemistry of living organisms?**
- A. Physiology**
  - B. Biochemistry**
  - C. Botany**
  - D. Entomology**
- 7. What is the chemical formula for water?**
- A. NaCl**
  - B. C<sub>2</sub>H<sub>6</sub>O**
  - C. H<sub>2</sub>O**
  - D. NaOH**
- 8. What branch of science studies the structure and composition of Earth and other planets?**
- A. Astronomy**
  - B. Physics**
  - C. Chemistry**
  - D. Geology**
- 9. What is the energy derived by chemical reactions in a living organism called?**
- A. Charge**
  - B. Surface area**
  - C. Compound**
  - D. Valence**
- 10. What is the difference between a solution and a suspension?**
- A. Suspensions settle, solutions do not**
  - B. Solutions settle, suspensions do not**
  - C. Solutions can be filtered, suspensions can not**
  - D. Suspensions can be filtered, solutions can not**

## **Answers**

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1. C
2. D
3. A
4. B
5. A
6. B
7. C
8. D
9. D
10. A

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## **Explanations**

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**1. What is the scientific name for the study of energy and matter?**

- A. Biology**
- B. Chemistry**
- C. Physics**
- D. Psychology**

Biological studies primarily focus on living organisms, while psychological studies focus on the mind and behavior. These two options are incorrect as they do not relate to the study of energy and matter. Chemistry is the study of the composition, properties, and reactions of substances, and while it does involve some aspects of energy and matter, it is a broader field that also encompasses the study of substances without a focus on energy and matter itself, making it an incorrect option. Physics, on the other hand, is the scientific study of energy, matter, and their interactions. Its principles are used to explain phenomena in fields such as chemistry and biology, making it the correct answer for the study of energy and matter.

**2. What is the study of the structure and behavior of systems made up of many interacting parts?**

- A. Ecology**
- B. Economics**
- C. Thermodynamics**
- D. Systems Theory**

Systems Theory is the study of the structure and behavior of systems made up of many interacting parts. Ecology is the study of how organisms interact with each other and their environment, and is not focused on the structure or behavior of systems. Economics is the study of production, distribution, and consumption of goods and services and does not focus on systems. Thermodynamics is the study of energy and its transformations, but it does not specifically focus on systems and their interactions. Therefore, the correct answer is D.

**3. What is the smallest representative unit in chemistry?**

- A. Atom**
- B. Molecule**
- C. Ion**
- D. Proton**

Atoms are the smallest representative unit in chemistry, as they are the basic building blocks of matter. A molecule is made up of two or more atoms bonded together, so it is not the smallest unit. An ion is an atom or molecule that has gained or lost electrons, making it electrically charged, so it is also not the smallest unit. A proton is a subatomic particle found in the nucleus of an atom, but it is only one component of an atom and not the entire unit itself. Therefore, the correct answer is A Atom.

**4. What is the acronym for the study of living organisms and their environment?**

**A. Biodiversity**

**B. Ecology**

**C. Earth Science**

**D. Environmental Science**

Ecology is the study of living organisms and their environment, while biodiversity specifically refers to the variety and distribution of life on Earth. Earth Science and Environmental Science are more umbrella terms that encompass a broader range of scientific studies, rather than being specific to living organisms and their environment.

**5. What is the most abundant element in the Earth's crust?**

**A. Oxygen**

**B. Iron**

**C. Carbon**

**D. Hydrogen**

Oxygen is the most abundant element in the Earth's crust, making up about 46.6% of its mass. This is because oxygen is a major component of the most common minerals found in the crust, such as silica and feldspar. Iron is the second most abundant element, making up about 27.7% of the crust's mass. While this is still a significant amount, it is much less than oxygen. Carbon is present in the Earth's crust, but in much lower amounts, typically less than 0.2% of the crust's mass. It is mainly found in carbonate minerals and organic matter. Hydrogen is not considered to be a major component of the Earth's crust, as it is a gas at normal crustal temperatures and does not form stable minerals.

**6. What is the study of the chemistry of living organisms?**

**A. Physiology**

**B. Biochemistry**

**C. Botany**

**D. Entomology**

Biochemistry is the study of the chemistry of living organisms, including their chemical structures and processes. Physiology (option A) is the study of the functions and activities of living organisms and their parts. Botany (option C) is the study of plants, while entomology (option D) is the study of insects. While all of these fields may overlap or be related to biochemistry, they specifically focus on different aspects of living organisms. Therefore, they are not the study of the chemistry of living organisms and are not the correct answer.

**7. What is the chemical formula for water?**

- A. NaCl
- B. C<sub>2</sub>H<sub>6</sub>O
- C. H<sub>2</sub>O**
- D. NaOH

Water is a compound composed of two hydrogen atoms and one oxygen atom, which is represented by the chemical formula H<sub>2</sub>O. This is in contrast to NaCl, which represents sodium chloride or table salt, C<sub>2</sub>H<sub>6</sub>O which represents ethanol or drinking alcohol, and NaOH which represents sodium hydroxide or commonly known as lye. These options do not accurately represent the chemical composition of pure water and are therefore incorrect.

**8. What branch of science studies the structure and composition of Earth and other planets?**

- A. Astronomy
- B. Physics
- C. Chemistry
- D. Geology**

Geology is the scientific study of the structure and composition of Earth and other planets. This includes studying the materials that make up the planet, the processes that shape it, and the history of its formation. Astronomy, on the other hand, focuses on the study of celestial objects such as stars, galaxies, and the universe as a whole. Physics is the science of matter, energy, and the interactions between them. Chemistry is the study of the composition, properties, and reactions of substances. While all of these branches of science may touch on aspects related to Earth and other planets, geology specifically focuses on studying their structure and composition. This makes D the most accurate answer to this question.

**9. What is the energy derived by chemical reactions in a living organism called?**

- A. Charge
- B. Surface area
- C. Compound
- D. Valence**

Valence is the term used to describe the difference between the inner and outer shells of an atom. The inner shell is made up of the innermost electrons, which have a lower energy level, while the outer shell contains the outermost electrons, which are at a higher energy level. The difference between these two energy levels is known as the atom's valence. Option A, charge, is incorrect because while the difference in charge between the two shells does play a role, it is not the defining factor. Option B, surface area, is also incorrect as it does not accurately describe the difference between the shells of an atom. Option C, compound, is incorrect because compounds are formed when atoms of different elements interact with each other, not based on the difference between their inner and outer shells. Overall, valence is the most appropriate term for the difference between the inner and outer shells of an atom.

**10. What is the difference between a solution and a suspension?**

**A. Suspensions settle, solutions do not**

**B. Solutions settle, suspensions do not**

**C. Solutions can be filtered, suspensions can not**

**D. Suspensions can be filtered, solutions can not**

Solutions and suspensions are two types of mixtures that are often compared to each other. A solution is a homogeneous mixture formed when a solute (usually a solid) is dissolved in a solvent (usually a liquid). On the other hand, a suspension is a heterogeneous mixture formed when small particles of a solid are dispersed throughout a liquid without dissolving. Option A is the correct answer because suspensions do indeed settle over time, as the solid particles are not dissolved and will eventually settle to the bottom. Solutions do not settle because the substances are dissolved and evenly dispersed throughout the mixture. Option B is incorrect because, as mentioned, solutions do not settle. Solutions have even distribution of particles and do not separate over time. Option C is also incorrect because solutions can be filtered, especially if the particles are large enough to be filtered out. Finally, option D is incorrect because



## Next Steps

**Congratulations on reaching the final section of this guide. You've taken a meaningful step toward passing your certification exam and advancing your career.**

**As you continue preparing, remember that consistent practice, review, and self-reflection are key to success. Make time to revisit difficult topics, simulate exam conditions, and track your progress along the way.**

**If you need help, have suggestions, or want to share feedback, we'd love to hear from you. Reach out to our team at [hello@examzify.com](mailto:hello@examzify.com).**

**Or visit your dedicated course page for more study tools and resources:**

**<https://naturalsciencescleppractice.examzify.com>**

**We wish you the very best on your exam journey. You've got this!**