

# CWEA Advanced Water Treatment Practice Test (Sample)

## Study Guide



**Everything you need from our exam experts!**

**Copyright © 2025 by Examzify - A Kaluba Technologies Inc. product.**

**ALL RIGHTS RESERVED.**

**No part of this book may be reproduced or transferred in any form or by any means, graphic, electronic, or mechanical, including photocopying, recording, web distribution, taping, or by any information storage retrieval system, without the written permission of the author.**

**Notice: Examzify makes every reasonable effort to obtain from reliable sources accurate, complete, and timely information about this product.**

**SAMPLE**

## **Questions**

- 1. What types of microorganisms are predominantly used in secondary treatment systems?**
  - A. Bacteria**
  - B. Viruses**
  - C. Protozoa**
  - D. Fungi**
- 2. What contaminant is specifically targeted in phosphorus removal practices?**
  - A. Nitrate**
  - B. Orthophosphate**
  - C. Ammonia**
  - D. Bacteria**
- 3. What percentage does 1-log removal correspond to?**
  - A. 90%**
  - B. 99%**
  - C. 99.9%**
  - D. 99.99%**
- 4. What does "granular activated carbon" (GAC) primarily remove from water?**
  - A. Heavy metals**
  - B. Microorganisms**
  - C. Organic contaminants, taste, and odor**
  - D. Inorganic compounds**
- 5. Which process involves the breakdown of organic material with the help of oxygen?**
  - A. Anaerobic digestion**
  - B. Aerobic digestion**
  - C. Chemical oxidation**
  - D. Filtration**

- 6. Which of the following is a method to reduce water hardness?**
- A. Reverse osmosis**
  - B. Chlorination**
  - C. Aerobic digestion**
  - D. Flocculation**
- 7. What is the main purpose of using a coagulant in water treatment?**
- A. To enhance flavor and odor**
  - B. To promote precipitation of contaminants**
  - C. To stabilize temperature variations**
  - D. To increase dissolved oxygen levels**
- 8. Which of the following is NOT typically a problem associated with secondary treatment?**
- A. Excessive foam generation**
  - B. Inadequate nutrient removal**
  - C. Overgrowth of algae**
  - D. Corrosion of pipes**
- 9. How does sedimentation work to improve water quality?**
- A. By adding coagulants**
  - B. By allowing heavier particles to settle to the bottom**
  - C. By increasing water temperature**
  - D. By aerating the water**
- 10. How does activated sludge contribute to biological treatment?**
- A. It isolates pathogens for elimination**
  - B. It enhances microbial growth and removal of pollutants**
  - C. It stabilizes water temperature**
  - D. It reduces sediment formation**

## **Answers**

SAMPLE

- 1. A**
- 2. B**
- 3. A**
- 4. C**
- 5. B**
- 6. A**
- 7. B**
- 8. D**
- 9. B**
- 10. B**

SAMPLE

## **Explanations**

**1. What types of microorganisms are predominantly used in secondary treatment systems?**

- A. Bacteria**
- B. Viruses**
- C. Protozoa**
- D. Fungi**

In secondary treatment systems, bacteria are the primary microorganisms used due to their essential role in the biological decomposition of organic matter present in wastewater. Bacteria are highly efficient at breaking down complex organic compounds through metabolic processes, resulting in the conversion of these compounds into simpler substances that can be further processed or safely released into the environment. Bacteria thrive in various environments present in secondary treatment systems, such as activated sludge processes and biological filtration. Their ability to reproduce quickly and adapt to different conditions makes them ideal for maintaining the required microbial populations for effective treatment. While viruses, protozoa, and fungi do play roles in certain environmental processes, they are not primarily responsible for organic matter degradation in secondary treatment systems. Viruses do not contribute to the breakdown of organic material but can affect bacterial populations. Protozoa, although they exist in these systems, mainly feed on bacteria and are less significant in terms of the overall treatment process. Fungi, meanwhile, are more commonly associated with the breakdown of organic materials in environments like soil or decaying matter rather than wastewater treatment.

**2. What contaminant is specifically targeted in phosphorus removal practices?**

- A. Nitrate**
- B. Orthophosphate**
- C. Ammonia**
- D. Bacteria**

Phosphorus removal practices primarily target orthophosphate as it constitutes the most bioavailable form of phosphorus in wastewater. This form is directly available for uptake by aquatic plants and algae, and its presence is a significant driver of eutrophication in water bodies, leading to detrimental environmental impacts such as harmful algal blooms. Decreasing orthophosphate concentrations is essential for the management of nutrient levels in treated water before it is discharged into natural water systems. Various treatment processes, such as chemical precipitation with agents like aluminum or iron salts, are employed specifically to reduce orthophosphate levels effectively. This is because orthophosphate contributes to excessive nutrient loading, which can degrade water quality and disrupt aquatic ecosystems. In contrast, nitrate, ammonia, and bacteria, while important in their own right for overall water quality, do not specifically pertain to the targeted phosphorus removal practices typically employed in wastewater treatment processes.

### 3. What percentage does 1-log removal correspond to?

- A. 90%**
- B. 99%**
- C. 99.9%**
- D. 99.99%**

1-log removal corresponds to a reduction of 90% of a particular microorganism or contaminant. In water treatment, the term "log removal" is used to express how effective a treatment process is at reducing the number of pathogens in water. Each "log" represents a tenfold reduction in concentration. Therefore, a 1-log reduction means that the concentration of the pathogen is reduced by a factor of 10, which equates to 90% removal of the original amount. This is crucial in water treatment processes where achieving high levels of pathogen removal is necessary for ensuring the safety and quality of drinking water. Higher log reductions correspond to increasingly stringent removal percentages: 2-log represents 99% removal, 3-log represents 99.9% removal, and so forth. Understanding these figures is essential for professionals in the field, as they guide the effectiveness of treatment methods used in various water treatment systems.

### 4. What does "granular activated carbon" (GAC) primarily remove from water?

- A. Heavy metals**
- B. Microorganisms**
- C. Organic contaminants, taste, and odor**
- D. Inorganic compounds**

Granular activated carbon (GAC) is primarily effective in the removal of organic contaminants, as well as taste and odor from water. This is due to its large surface area and porous structure, which provides ample space for adsorption processes. When water passes through GAC, organic molecules attach to the carbon's surface, effectively reducing their concentration in the water. Organic contaminants often include substances like volatile organic compounds (VOCs), pesticides, and many other man-made chemicals that may affect the quality and safety of water. Taste and odor issues, frequently a result of these organic compounds or byproducts from disinfection processes (like chlorine), are also addressed effectively by GAC. While GAC can have some effectiveness against certain microorganisms, it's primarily aimed at organic substances. It is less effective in removing heavy metals or inorganic compounds, which tend to require different treatment methods, such as ion exchange or reverse osmosis, for effective removal. This specialization in addressing organic pollutants and associated taste and odor makes GAC a popular choice in water treatment systems.

**5. Which process involves the breakdown of organic material with the help of oxygen?**

- A. Anaerobic digestion**
- B. Aerobic digestion**
- C. Chemical oxidation**
- D. Filtration**

The process involving the breakdown of organic material with the help of oxygen is aerobic digestion. This biological process utilizes microorganisms that thrive in the presence of oxygen to decompose organic matter, converting it into stable, non-toxic products. Aerobic digestion primarily aims to reduce the volume of waste sludge and stabilize organic materials, making it a crucial step in wastewater treatment. During aerobic digestion, the organic material is oxidized, resulting in the production of carbon dioxide, water, and microbial biomass. This method is efficient for treating various organic wastes and can significantly reduce the amount of sludge generated, thus minimizing disposal costs and impacts on the environment. In contrast, anaerobic digestion occurs in environments devoid of oxygen, leading to the production of biogas (mainly methane and carbon dioxide) along with digested organic material. Chemical oxidation involves the use of chemicals, rather than biological processes, to break down contaminants. Filtration is a physical separation process used to remove solids from liquids but does not inherently involve the breakdown of organic material with oxygen.

**6. Which of the following is a method to reduce water hardness?**

- A. Reverse osmosis**
- B. Chlorination**
- C. Aerobic digestion**
- D. Flocculation**

Reverse osmosis is an effective method to reduce water hardness because it uses a semi-permeable membrane to separate ions and larger particles from water. This process is particularly capable of removing dissolved minerals such as calcium and magnesium, which are the primary contributors to water hardness. As water is forced through the membrane, these hardness ions are left behind, resulting in softened water on the other side. The other methods mentioned are not primarily aimed at reducing water hardness. Chlorination is used for disinfection and does not significantly impact hardness. Aerobic digestion is a biological process for treating organic waste and does not deal directly with hardness. Flocculation involves the aggregation of particles to aid in their removal during water treatment but does not specifically target hardness-causing minerals.

**7. What is the main purpose of using a coagulant in water treatment?**

- A. To enhance flavor and odor**
- B. To promote precipitation of contaminants**
- C. To stabilize temperature variations**
- D. To increase dissolved oxygen levels**

The primary purpose of using a coagulant in water treatment is to promote the precipitation of contaminants. Coagulants work by neutralizing the charges on suspended particles, allowing them to clump together into larger aggregates, or flocs. This process makes it easier to remove these aggregates from the water during subsequent filtration or sedimentation steps. By enhancing the removal of turbidity and other pollutants, coagulants play a crucial role in producing clean and safe drinking water. The other options focus on aspects that are not directly related to the coagulation process in water treatment. For instance, enhancing flavor and odor is not the main role of coagulants, as they primarily target physical contaminants rather than taste or smell. Additionally, stabilizing temperature variations does not relate to the coagulation process, and increasing dissolved oxygen levels is typically addressed through aeration rather than the use of coagulants. Thus, option B accurately reflects the function of coagulants in water treatment operations.

**8. Which of the following is NOT typically a problem associated with secondary treatment?**

- A. Excessive foam generation**
- B. Inadequate nutrient removal**
- C. Overgrowth of algae**
- D. Corrosion of pipes**

Secondary treatment in wastewater treatment is primarily focused on the biological processes that remove organic matter and nutrients from effluent. Typically, the problems associated with secondary treatment often involve the biological components and processes. Excessive foam generation can occur due to the presence of surfactants or excessive biological activity, which can hinder the efficiency of treatment operations. Inadequate nutrient removal relates to the challenges of sufficiently reducing nitrogen and phosphorus levels, which can be especially problematic if the secondary treatment system is not properly designed or operated. Overgrowth of algae can result from nutrient-rich effluent being discharged into receiving waters, leading to eutrophication and other ecological issues. Corrosion of pipes, on the other hand, is generally not a direct concern of secondary treatment processes. Pipe corrosion is more associated with factors such as the chemical composition of the effluent, the materials used in the construction of the piping, and the overall system design, rather than the biological processes occurring during the secondary treatment stage. Therefore, while it is a concern in wastewater systems, it does not specifically stem from issues related to secondary treatment.

## 9. How does sedimentation work to improve water quality?

- A. By adding coagulants
- B. By allowing heavier particles to settle to the bottom**
- C. By increasing water temperature
- D. By aerating the water

Sedimentation is a critical process in water treatment that enhances water quality primarily by allowing heavier particles, such as sediments, silt, and other suspended solids, to settle to the bottom of a treatment tank or basin. This sedimentation occurs due to the force of gravity acting on the particles, causing them to gradually fall out of suspension within the water column. As these heavier particles settle, the water above becomes clearer and contains fewer impurities. This reduction in solid material is essential in preparing the water for subsequent treatment processes, including filtration and disinfection. By removing suspended solids, sedimentation effectively decreases the turbidity of the water and can significantly reduce the organic load, making it easier for later treatment stages to provide safe drinking water. The other options present methods that are not directly related to the sedimentation process. Adding coagulants (the first choice) is a preceding step that prepares the water for sedimentation, while increasing water temperature (the third option) and aerating the water (the fourth option) are involved in different treatment processes that do not directly relate to the settling of particles in sedimentation.

## 10. How does activated sludge contribute to biological treatment?

- A. It isolates pathogens for elimination
- B. It enhances microbial growth and removal of pollutants**
- C. It stabilizes water temperature
- D. It reduces sediment formation

Activated sludge plays a crucial role in biological treatment processes primarily by enhancing microbial growth and facilitating the removal of pollutants from wastewater. In a typical activated sludge system, aeration tanks are used to mix wastewater with a concentrated population of microorganisms, primarily bacteria. These microorganisms consume organic matter and other pollutants in the wastewater, breaking them down into simpler, less harmful substances. The process relies on a well-balanced environment for the microbes to thrive, which is supported by the presence of activated sludge. More specifically, the activated sludge forms flocs, which are clusters of bacteria and other microorganisms. These flocs not only provide a larger surface area for microbial activity but also help in the efficient settling of solids afterward, allowing for the separation of treated water from leftover sludge. This process significantly improves the overall efficiency of wastewater treatment, as the enhanced microbial growth leads to a more effective breakdown of contaminants, including organic matter, nutrients, and pathogens, thereby producing cleaner effluent. As a result, the biological treatment via activated sludge is integral to meeting regulatory discharge standards and protecting environmental water quality.