

# Biochemistry Module 6 Practice Exam (Sample)

## Study Guide



**Everything you need from our exam experts!**

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# Introduction

Preparing for a certification exam can feel overwhelming, but with the right tools, it becomes an opportunity to build confidence, sharpen your skills, and move one step closer to your goals. At Examzify, we believe that effective exam preparation isn't just about memorization, it's about understanding the material, identifying knowledge gaps, and building the test-taking strategies that lead to success.

This guide was designed to help you do exactly that.

Whether you're preparing for a licensing exam, professional certification, or entry-level qualification, this book offers structured practice to reinforce key concepts. You'll find a wide range of multiple-choice questions, each followed by clear explanations to help you understand not just the right answer, but why it's correct.

The content in this guide is based on real-world exam objectives and aligned with the types of questions and topics commonly found on official tests. It's ideal for learners who want to:

- Practice answering questions under realistic conditions,
- Improve accuracy and speed,
- Review explanations to strengthen weak areas, and
- Approach the exam with greater confidence.

We recommend using this book not as a stand-alone study tool, but alongside other resources like flashcards, textbooks, or hands-on training. For best results, we recommend working through each question, reflecting on the explanation provided, and revisiting the topics that challenge you most.

**Remember:** successful test preparation isn't about getting every question right the first time, it's about learning from your mistakes and improving over time. Stay focused, trust the process, and know that every page you turn brings you closer to success.

Let's begin.

# How to Use This Guide

**This guide is designed to help you study more effectively and approach your exam with confidence. Whether you're reviewing for the first time or doing a final refresh, here's how to get the most out of your Examzify study guide:**

## **1. Start with a Diagnostic Review**

**Skim through the questions to get a sense of what you know and what you need to focus on. Your goal is to identify knowledge gaps early.**

## **2. Study in Short, Focused Sessions**

**Break your study time into manageable blocks (e.g. 30 - 45 minutes). Review a handful of questions, reflect on the explanations.**

## **3. Learn from the Explanations**

**After answering a question, always read the explanation, even if you got it right. It reinforces key points, corrects misunderstandings, and teaches subtle distinctions between similar answers.**

## **4. Track Your Progress**

**Use bookmarks or notes (if reading digitally) to mark difficult questions. Revisit these regularly and track improvements over time.**

## **5. Simulate the Real Exam**

**Once you're comfortable, try taking a full set of questions without pausing. Set a timer and simulate test-day conditions to build confidence and time management skills.**

## **6. Repeat and Review**

**Don't just study once, repetition builds retention. Re-attempt questions after a few days and revisit explanations to reinforce learning. Pair this guide with other Examzify tools like flashcards, and digital practice tests to strengthen your preparation across formats.**

**There's no single right way to study, but consistent, thoughtful effort always wins. Use this guide flexibly, adapt the tips above to fit your pace and learning style. You've got this!**

## Questions

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- 1. What does heterocyclic mean?**
  - A. A ring-shaped carbon chain that is entirely carbon**
  - B. A ring-shaped carbon chain that has at least 1 atom that isn't carbon in the ring**
  - C. A straight chain with alternating double bonds**
  - D. A ring with nitrogen only**
- 2. Which base has a methyl group and is thus characteristic of DNA?**
  - A. Uracil**
  - B. Thymine**
  - C. Cytosine**
  - D. Adenine**
- 3. What is the complementary strand of DNA for the base sequence 5'-GCATGAGCC-3'?**
  - A. 3'-CGTACTCGG-5'**
  - B. 3'-GCTATCGGA-5'**
  - C. 3'-CGTACGCTG-5'**
  - D. 3'-CGTACGCGG-5'**
- 4. Purines are characterized by having what kind of ring structure?**
  - A. One ring**
  - B. Three rings**
  - C. Two rings**
  - D. None**
- 5. Which base pairs with cytosine in Watson-Crick base pairing?**
  - A. Adenine**
  - B. Cytosine**
  - C. Uracil**
  - D. Guanine**

- 6. Which of the following bases is a purine?**
- A. Uracil**
  - B. Guanine**
  - C. Thymine**
  - D. Cytosine**
- 7. What percentage of ribosomes' mass is composed of rRNA?**
- A. 60%**
  - B. 65%**
  - C. 30%**
  - D. 50%**
- 8. Two DNA strands are described as complementary. What does this mean?**
- A. Adenine pairs with thymine and cytosine pairs with guanine.**
  - B. Each strand has identical sequence to the other.**
  - C. The sugar-phosphate backbones align in opposite directions.**
  - D. Bases form covalent bonds between strands**
- 9. The pentose sugar in RNA is ribose.**
- A. Glucose**
  - B. Deoxyribose**
  - C. Fructose**
  - D. Ribose**
- 10. In enzyme-catalyzed reactions, what role does FAD play?**
- A. Shuttle hydrogen between molecules**
  - B. Serve as a detergent**
  - C. Provide phosphate groups to substrates**
  - D. Bind to DNA to regulate transcription**

## Answers

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1. B
2. B
3. A
4. C
5. D
6. B
7. B
8. A
9. D
10. A

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## **Explanations**

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### 1. What does heterocyclic mean?

- A. A ring-shaped carbon chain that is entirely carbon
- B. A ring-shaped carbon chain that has at least 1 atom that isn't carbon in the ring**
- C. A straight chain with alternating double bonds
- D. A ring with nitrogen only

Heterocyclic refers to a ring structure that contains at least one atom other than carbon within the ring. That heteroatom is usually nitrogen, oxygen, or sulfur, and its presence changes the ring's properties like electron distribution and reactivity. So a ring that is all carbon is carbocyclic, and a straight chain with conjugated double bonds isn't a ring at all. A ring that includes a non-carbon atom fits the definition, which is why the statement describing a ring with at least one non-carbon atom in the ring is the correct way to define heterocycles. Examples you'll see often are pyridine (N in a six-member ring), furan (O in a five-member ring), and thiophene (S in a five-member ring).

### 2. Which base has a methyl group and is thus characteristic of DNA?

- A. Uracil
- B. Thymine**
- C. Cytosine
- D. Adenine

A methyl group on the base distinguishes DNA from RNA. Thymine, the DNA base, has a methyl group attached at the 5-position of its pyrimidine ring. Uracil, which is used in RNA, lacks this methyl group, and cytosine and adenine don't carry this methyl modification in their standard forms either. This methylated base helps mark DNA as the chromosome's blueprint, contributing to its stability and identity. So thymine is the base with a methyl group and is characteristic of DNA.

### 3. What is the complementary strand of DNA for the base sequence 5'-GCATGAGCC-3'?

- A. 3'-CGTACTCGG-5'**
- B. 3'-GCTATCGGA-5'
- C. 3'-CGTACGCTG-5'
- D. 3'-CGTACGCGG-5'

DNA bases pair specifically and the two strands run in opposite directions, so the complementary strand is built by swapping each base with its partner: G pairs with C, C with G, A with T, and T with A. For 5'-GCATGAGCC-3', replace each base accordingly and read the new strand from its 3' end to its 5' end: 3'-CGTACTCGG-5'. This arrangement preserves both the correct base pairs and the antiparallel orientation, making it the correct complementary strand.

**4. Purines are characterized by having what kind of ring structure?**

- A. One ring
- B. Three rings
- C. Two rings**
- D. None

Purines are bicyclic, built from two rings fused together: a five-membered imidazole ring attached to a six-membered pyrimidine ring. This two-ring system is the defining feature of purines, with examples like adenine and guanine. In contrast, pyrimidines have a single ring, so the statement that purines contain two rings is the correct descriptor. The fused rings form a single, planar aromatic system, which helps explain why purines pair with pyrimidines to maintain a consistent base-pairing width in DNA and RNA.

**5. Which base pairs with cytosine in Watson-Crick base pairing?**

- A. Adenine
- B. Cytosine
- C. Uracil
- D. Guanine**

Watson-Crick base pairing relies on complementary hydrogen-bonding patterns between a purine and a pyrimidine to keep the DNA double helix uniform in width. Cytosine, a pyrimidine, pairs with guanine, a purine, through three hydrogen bonds. This specific pairing provides the strongest and most stable interaction for that combination and preserves the structural geometry of the helix. In contrast, cytosine does not pair with itself, and adenine pairs with thymine (or uracil in RNA), not with cytosine. So the partner for cytosine in this canonical pairing is guanine.

**6. Which of the following bases is a purine?**

- A. Uracil
- B. Guanine**
- C. Thymine
- D. Cytosine

Purines are the two-ring nucleobases. Among the bases, adenine and guanine are purines, while cytosine, thymine, and uracil are pyrimidines (which have a single ring). Since guanine has the two-ring structure, it is the purine in this set. In DNA, purines pair with pyrimidines (guanine pairs with cytosine via three hydrogen bonds; adenine with thymine via two), helping maintain the uniform width of the helix. In RNA, thymine is replaced by uracil, but both thymine and uracil are pyrimidines.

**7. What percentage of ribosomes' mass is composed of rRNA?**

- A. 60%
- B. 65%**
- C. 30%
- D. 50%

Ribosomes are ribonucleoprotein complexes, meaning RNA and protein both contribute to their structure and function, but the RNA portion forms the bulk of the mass. The rRNA molecules create the scaffold and the catalytic center, so they make up roughly two-thirds of the ribosome's mass, with ribosomal proteins contributing the remaining one-third. In both bacteria and eukaryotes, this translates to about 60-65% of the mass being rRNA, which is why the commonly cited figure is around 65%.

**8. Two DNA strands are described as complementary. What does this mean?**

- A. Adenine pairs with thymine and cytosine pairs with guanine.**
- B. Each strand has identical sequence to the other.
- C. The sugar-phosphate backbones align in opposite directions.
- D. Bases form covalent bonds between strands

Complementary DNA strands have base sequences that fit together through specific base pairing: adenine pairs with thymine and cytosine pairs with guanine. This means the sequence on one strand determines the sequence on the other, so the two strands are not identical but complementary. The strands are held together by hydrogen bonds between these base pairs, while the sugar-phosphate backbones form the covalent links within each strand and run in opposite directions (antiparallel). This precise pairing arrangement is what makes the two strands able to come together to form the double helix and to serve as accurate templates during replication.

**9. The pentose sugar in RNA is ribose.**

- A. Glucose
- B. Deoxyribose
- C. Fructose
- D. Ribose**

RNA uses a five-carbon sugar, ribose, as its backbone component. This pentose has a hydroxyl group at the 2' carbon, which distinguishes it from DNA's sugar, deoxyribose, that lacks the 2' OH. The other options are hexoses (glucose and fructose) and do not serve as the sugar in RNA. Therefore, the pentose sugar in RNA is ribose.

**10. In enzyme-catalyzed reactions, what role does FAD play?**

- A. Shuttle hydrogen between molecules**
- B. Serve as a detergent**
- C. Provide phosphate groups to substrates**
- D. Bind to DNA to regulate transcription**

FAD acts as a redox cofactor that shuttles electrons (and associated protons) between molecules in enzyme-catalyzed reactions. In many dehydrogenases, the substrate donates two electrons and two protons to the oxidized FAD, forming FADH<sub>2</sub>. The enzyme then passes those electrons to the next carrier (for example, ubiquinone), regenerating FAD. This electron/hydrogen transfer role is why FAD is described as a carrier within the reaction. It's not a detergent, it doesn't donate phosphate groups to substrates, and it doesn't bind DNA to regulate transcription.

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## Next Steps

**Congratulations on reaching the final section of this guide. You've taken a meaningful step toward passing your certification exam and advancing your career.**

**As you continue preparing, remember that consistent practice, review, and self-reflection are key to success. Make time to revisit difficult topics, simulate exam conditions, and track your progress along the way.**

**If you need help, have suggestions, or want to share feedback, we'd love to hear from you. Reach out to our team at [hello@examzify.com](mailto:hello@examzify.com).**

**Or visit your dedicated course page for more study tools and resources:**

**<https://biochemmodule6.examzify.com>**

**We wish you the very best on your exam journey. You've got this!**

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